

Heterogeneous Atmospheric Dust

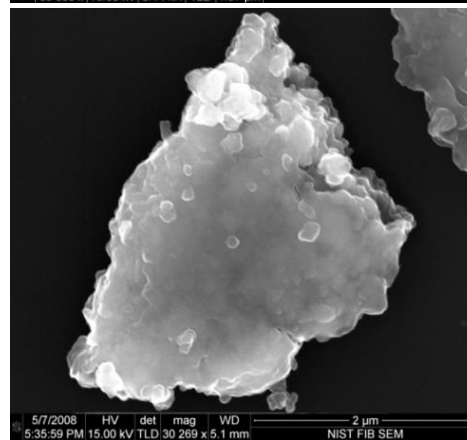
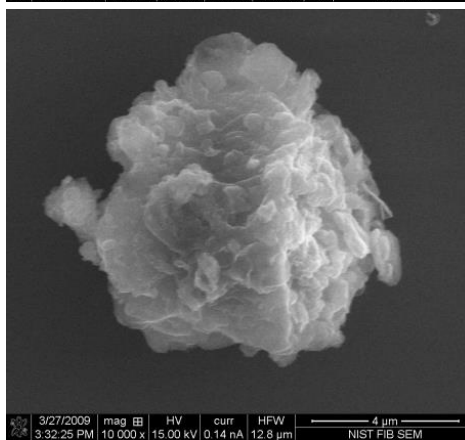
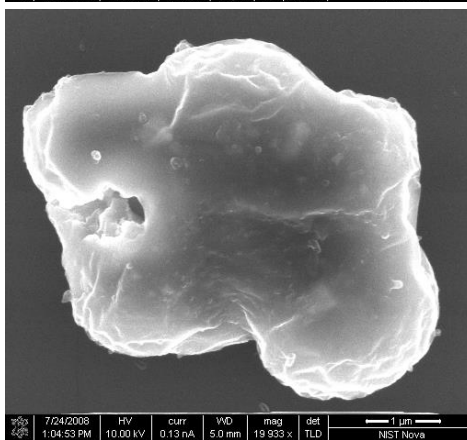
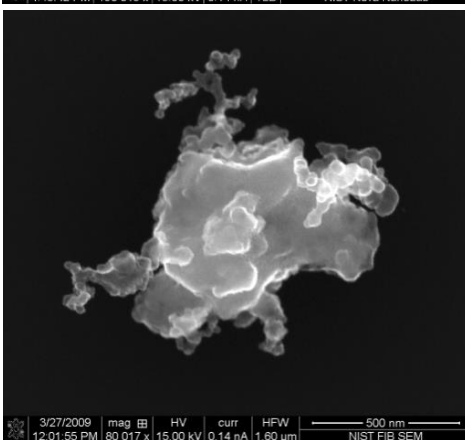
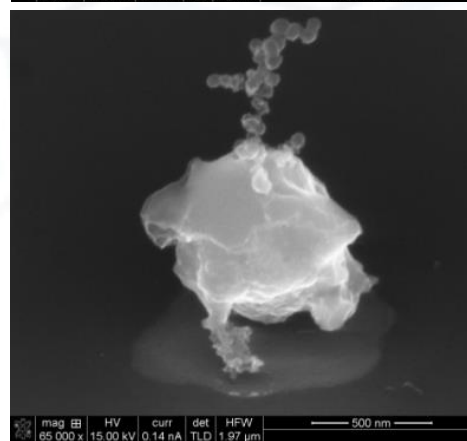
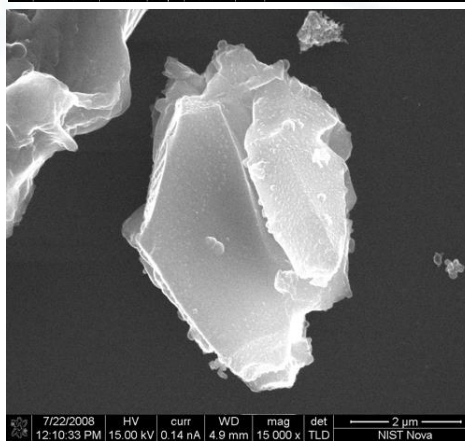
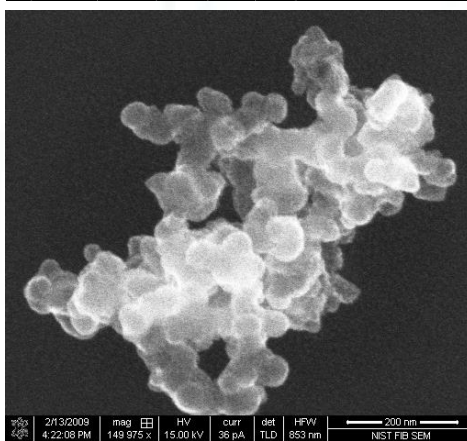
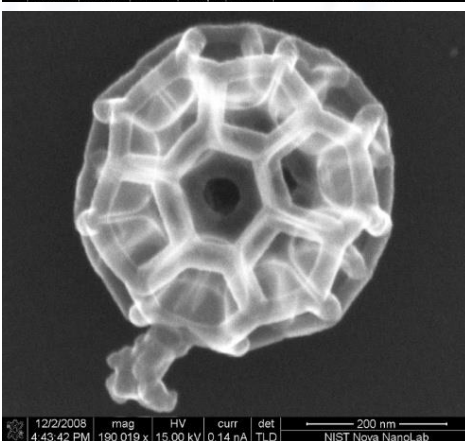
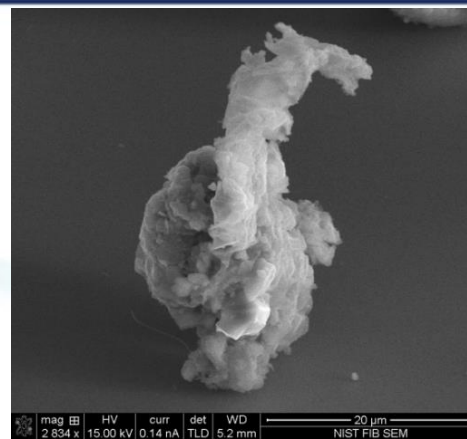
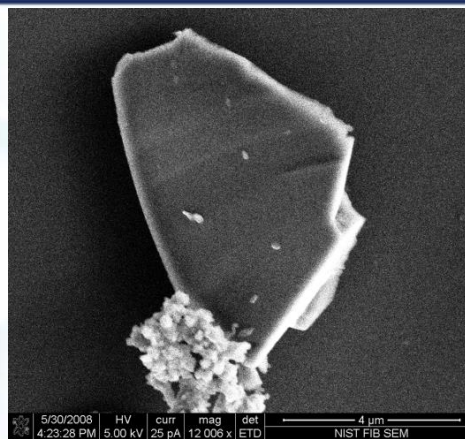
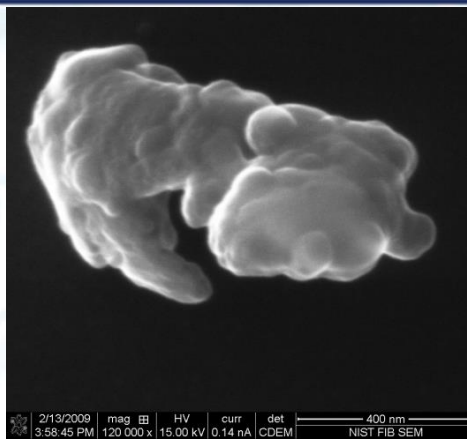
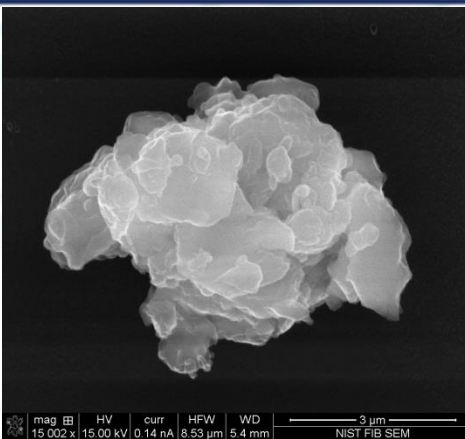
Individual Particle Microanalysis
Optical Property Modeling
Lab-Generated Particle Analogs

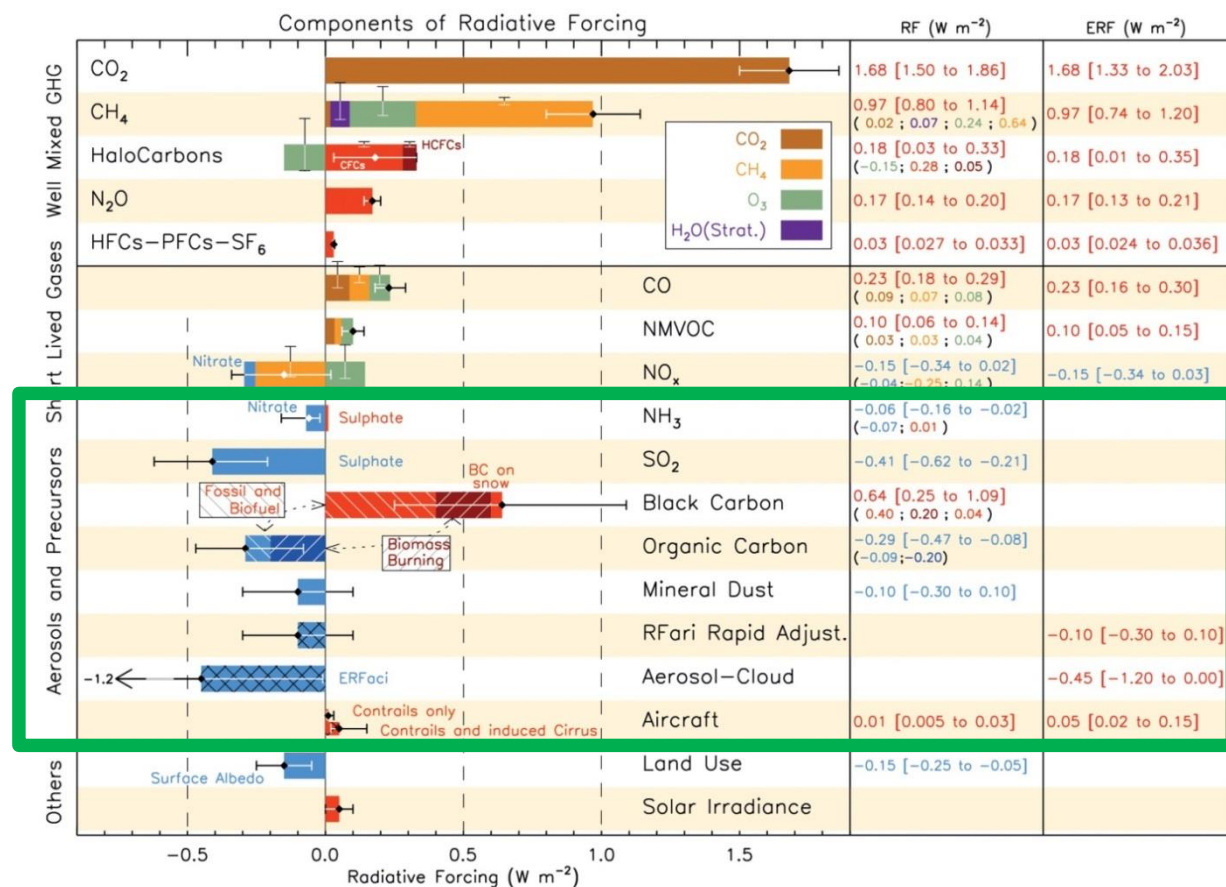
Joseph Conny
Materials Measurement Science Division

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	Sean M. Collins	Xiaofei Ma
	L. Reid Gunn	Robert D. Willis (EPA)

Overview

- Heterogeneous atmospheric dust:
 - Why important in climate change science
 - Goals of this research
- Microanalysis of individual particles:
 - Particle size, shape, chemical composition, internal structure
- Particle 3-D spatial models from focused ion-beam tomography
- Optical property modeling (absorption and scattering) based on 3-D spatial models
- Lab-generated atmospheric particle analogs with mixed optical phases
 - Paradigm for creating suite of mixed-phase reference materials



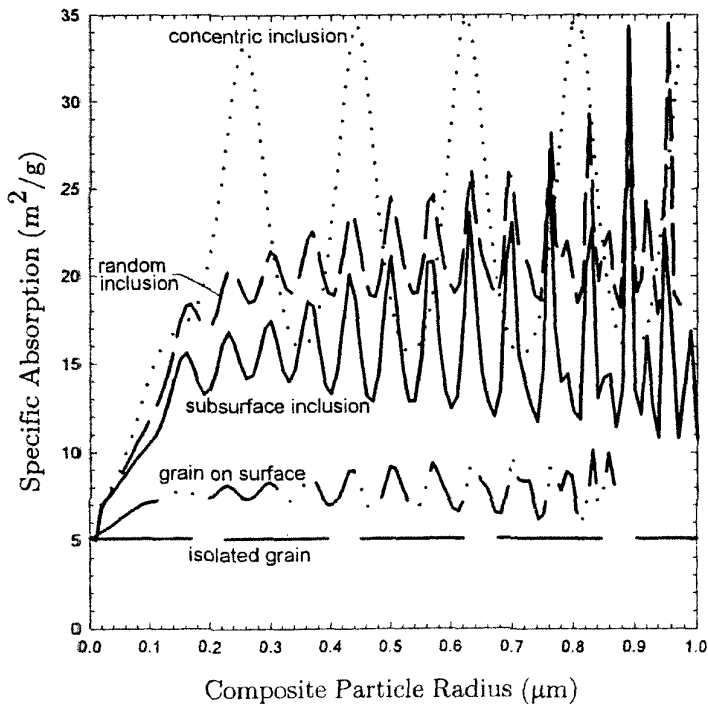


IPCC AR5, The Physical Science Basis, 2013

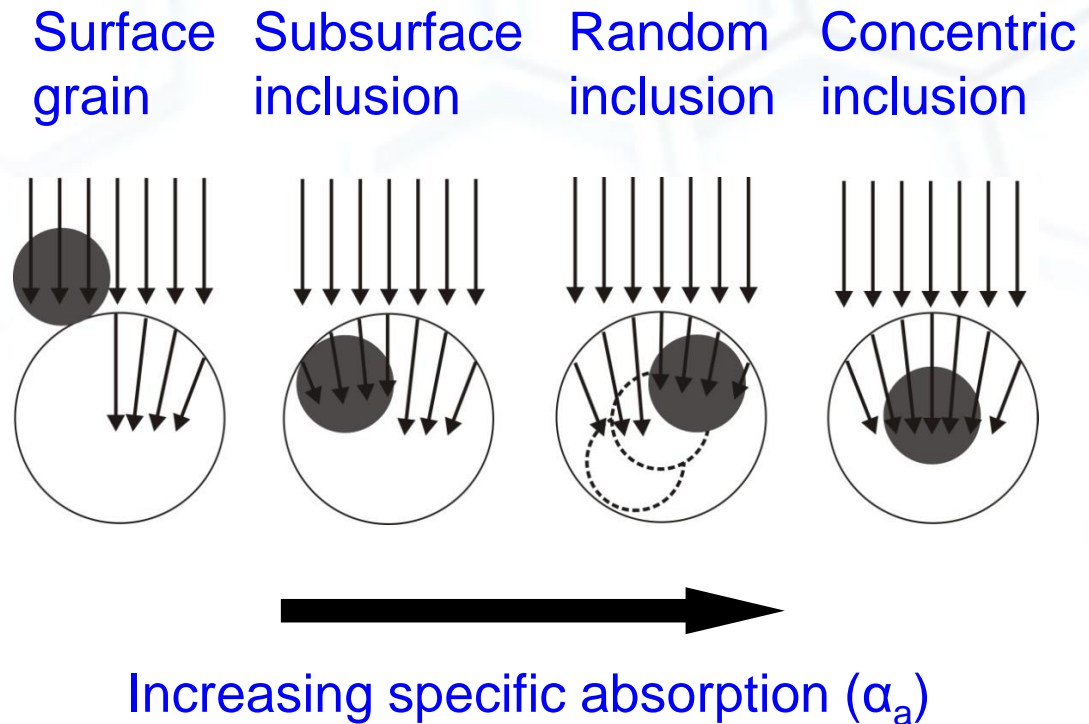
What's missing:

- Urban dust
- Aged aerosols (e.g., Asian pollution and dust transported to Hawaii)
- → Heterogeneous aerosols

Light Absorption in Internally-Mixed Particles



Fuller et al., *J. Geophys. Res.* 104
(1999) 15,941-15,954 (Figure 12)



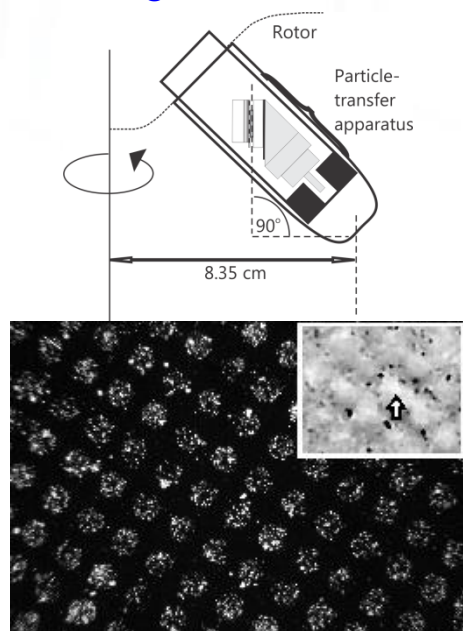
Research Goals

- **Determine the extent of natural variation in optical properties of heterogeneous particles (scattering and absorption efficiencies) based on optical models of selected individual particles**
- **Determine differences in optical properties with respect to**
 - Season
 - Geographic region (U.S. East, West, South)
 - Development (urban vs. rural)
 - Continental air vs marine air
- **Generate optically heterogeneous particles in the lab as analogs of real heterogeneous particles**

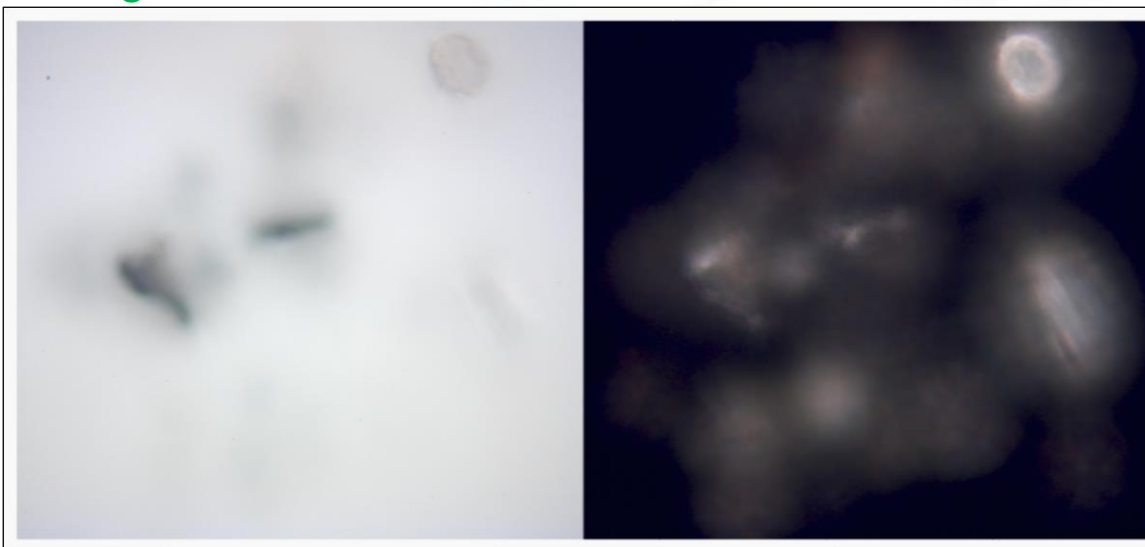
Microanalysis Process

- Sample preparation
- Survey of optical behavior → Light microscopy
- Particle population analysis → Automated SEM
- Individual particle SEM analysis → X-ray spectroscopy, electron backscatter diffraction

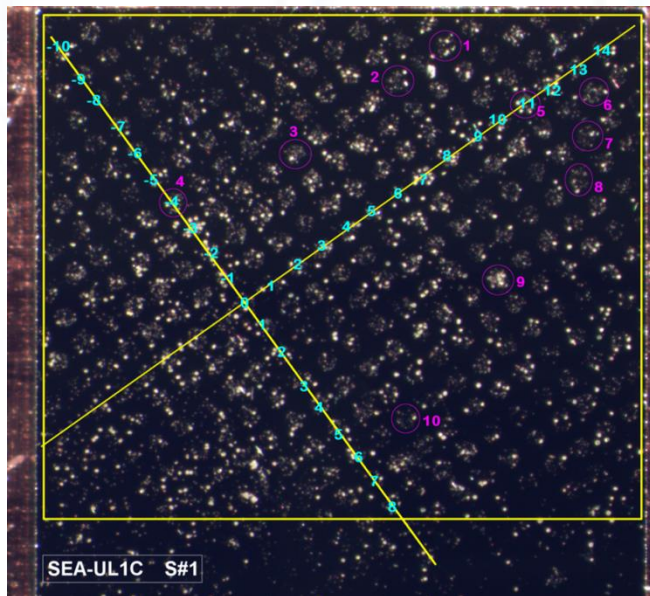
Particle Transfer:
Filter to germanium wafer



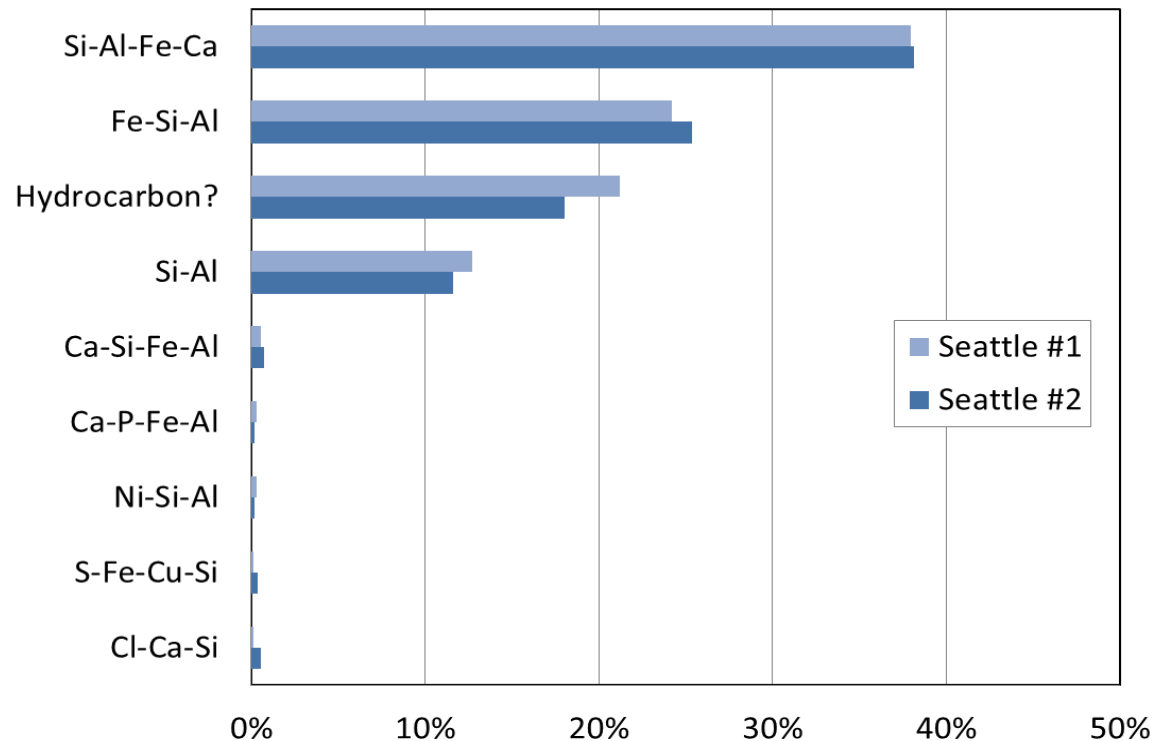
Light Microscopy: Seattle Winter 2005 Sample
Bright Field **Dark Field**



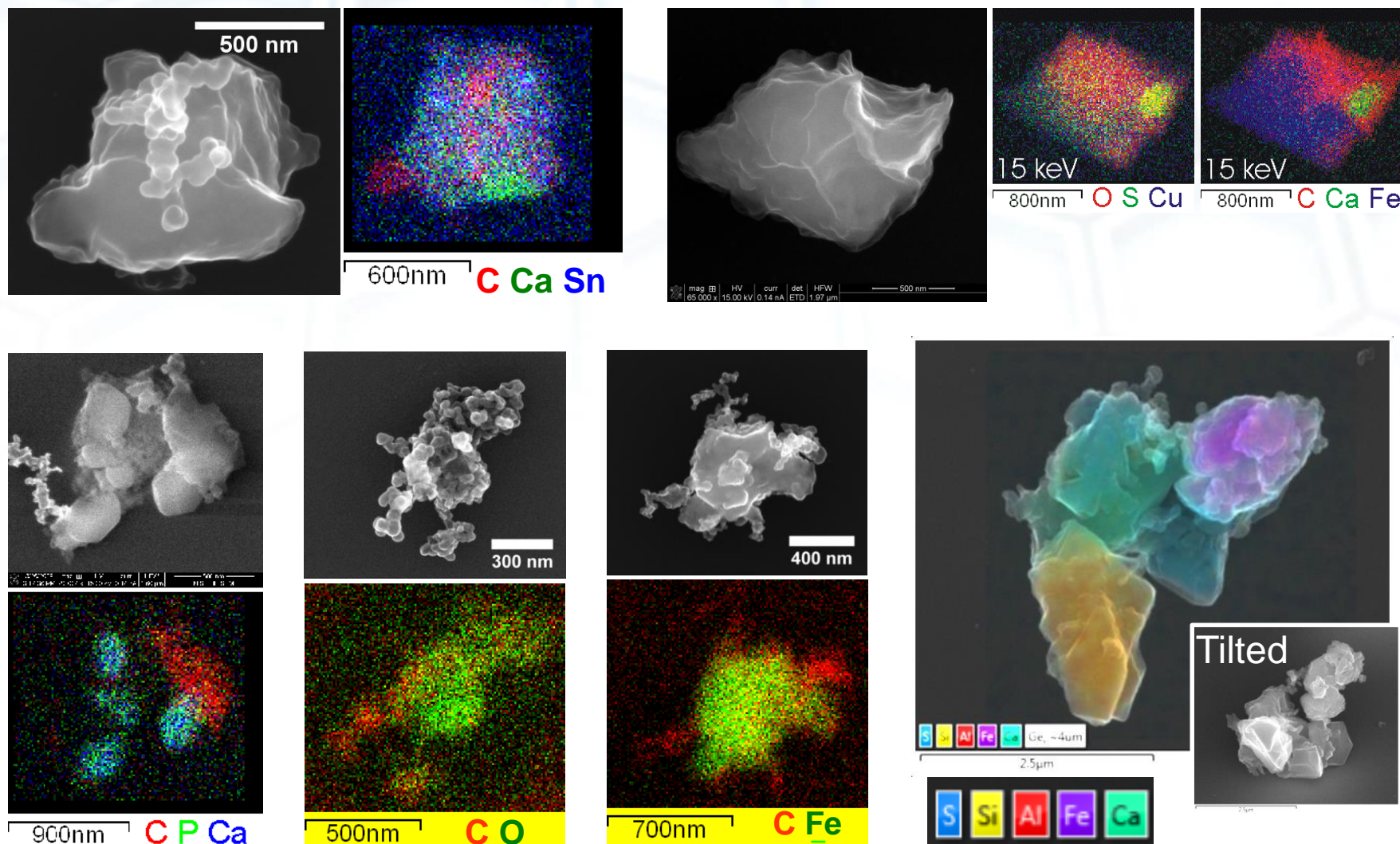
Analysis of Particle Populations



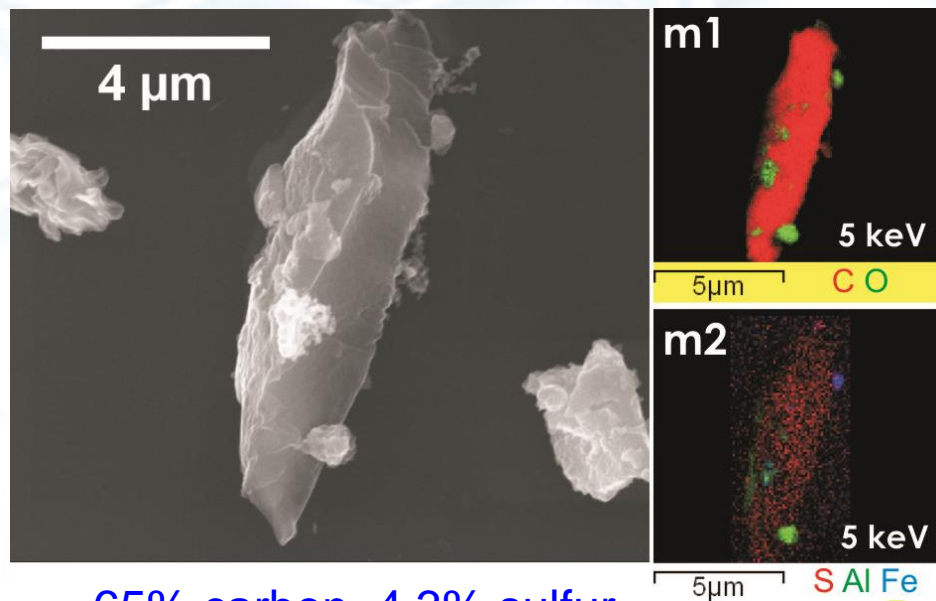
Seattle samples: Most prominent classes



Complexity of Urban Particles: Natural or Anthropogenic?



Optical Behavior of an Individual Particle

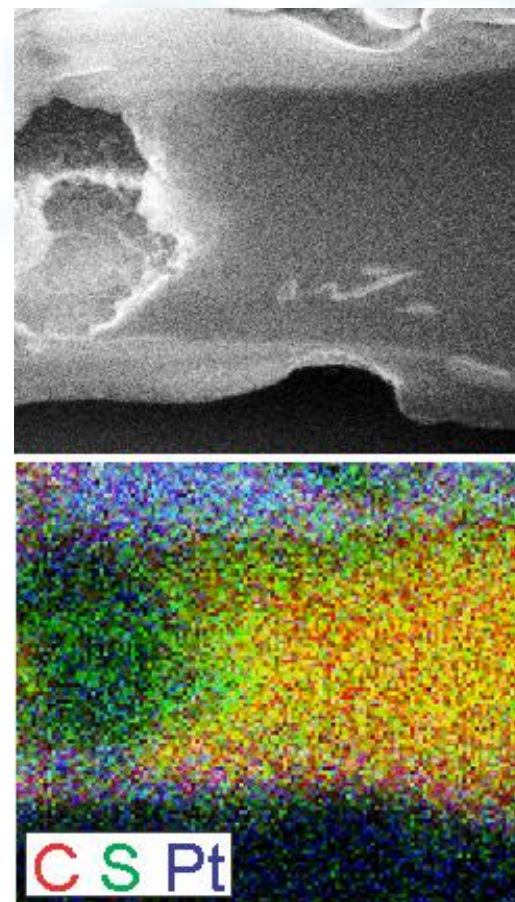


65% carbon, 4.3% sulfur

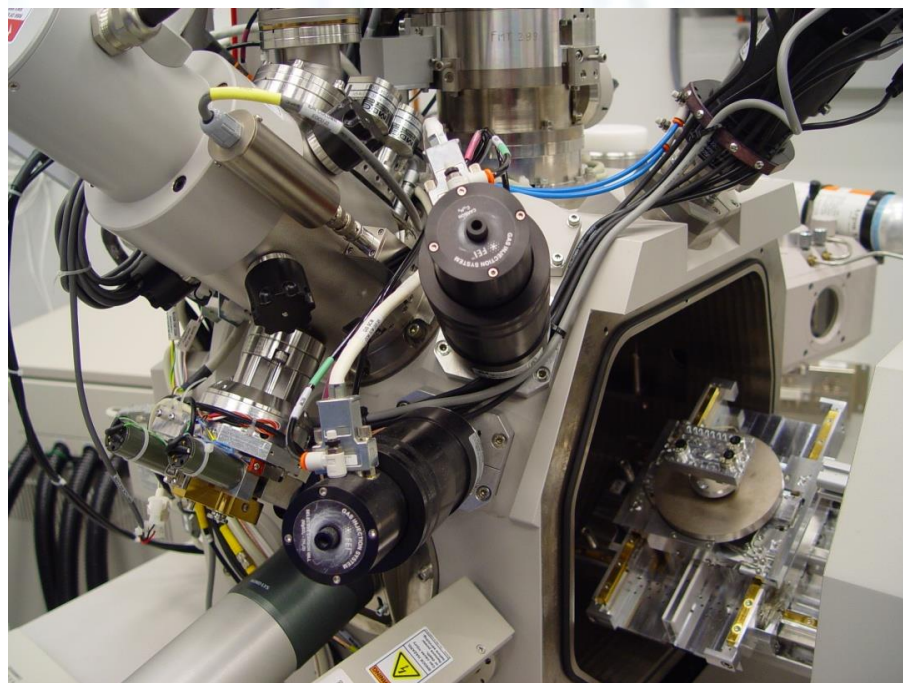
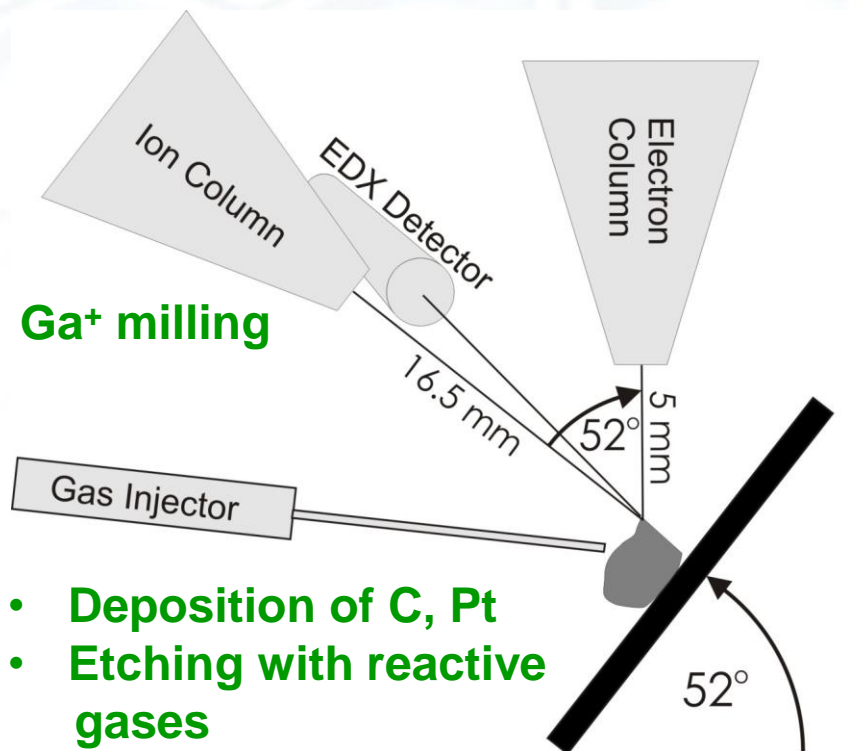
Is sulfur on the surface, e.g. $(\text{NH}_4)_2\text{SO}_4$, or throughout particle as in tire wear?

Single-scattering albedo (scattering/extinction ratio)

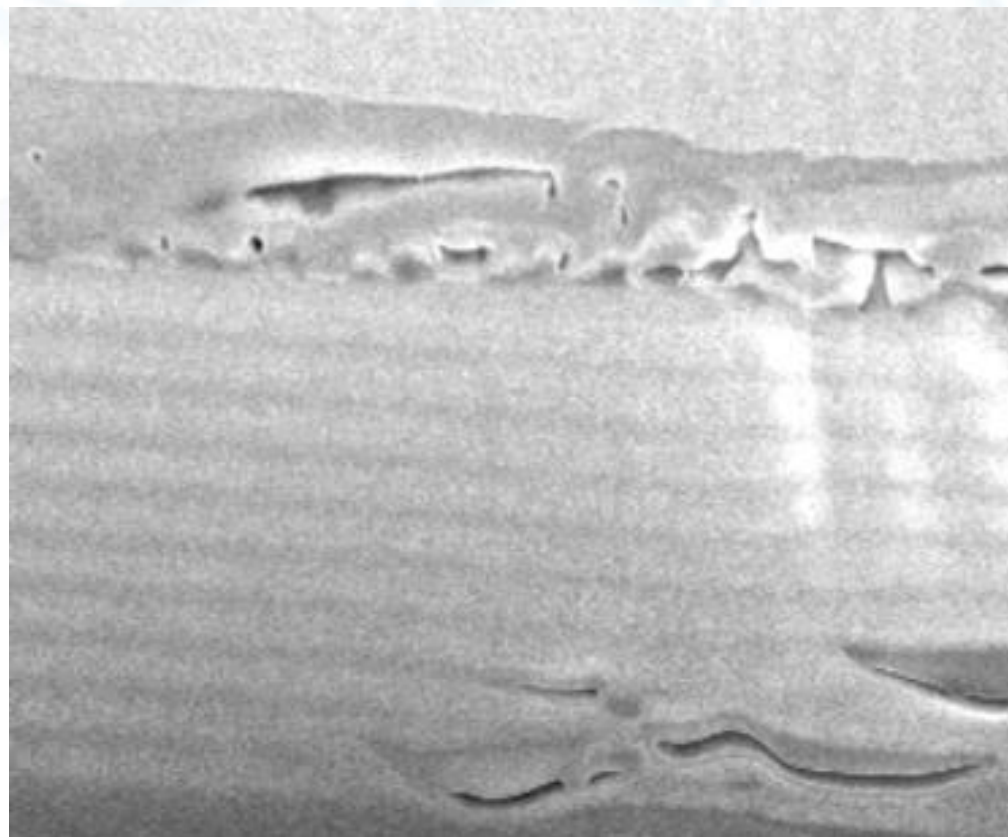
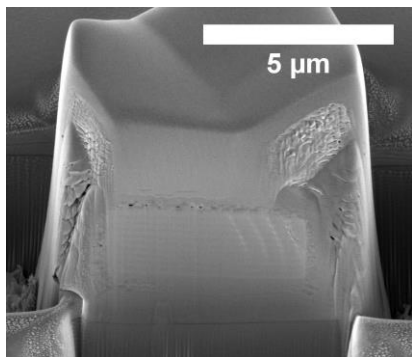
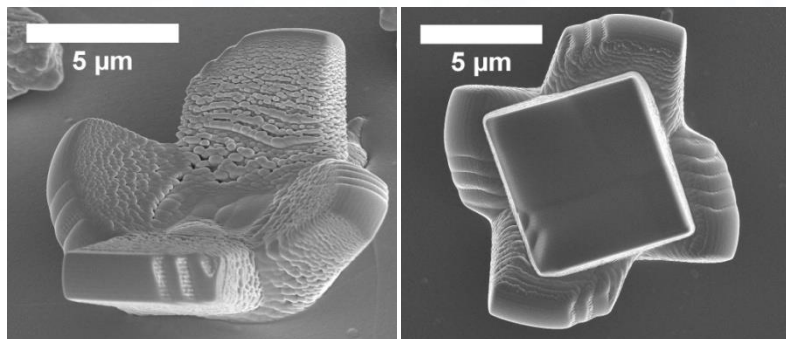
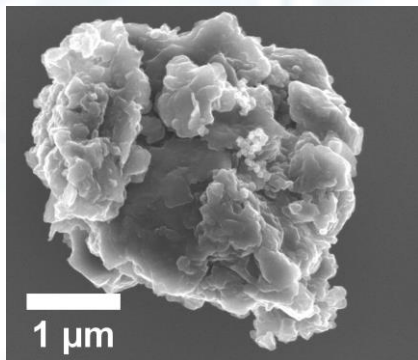
- Tire wear particle: 0.54
- If assume sulfate shell/soot core: 0.49
- If assume sulfate shell/graphite core: 0.54
 - But, backscattering 37% higher than tire wear!



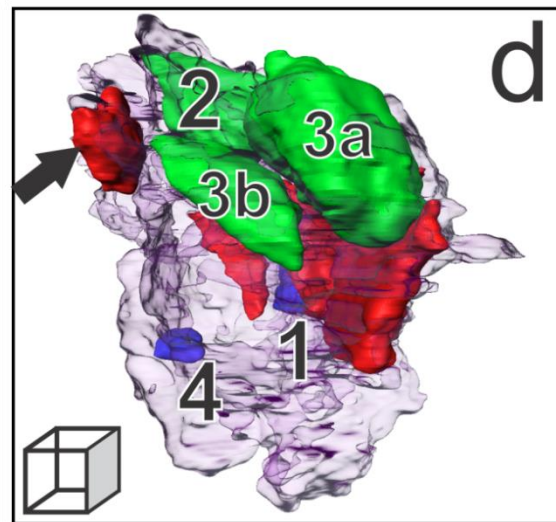
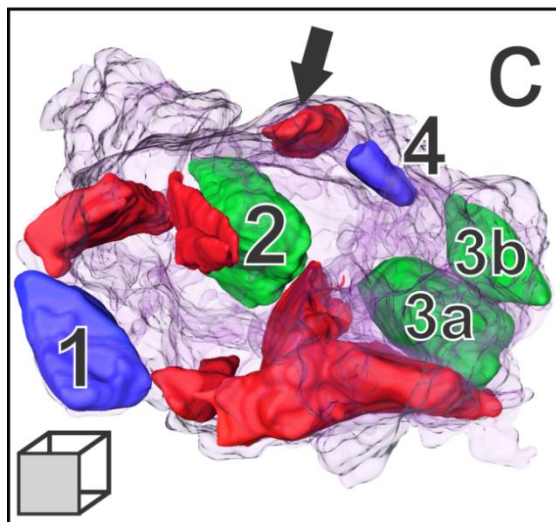
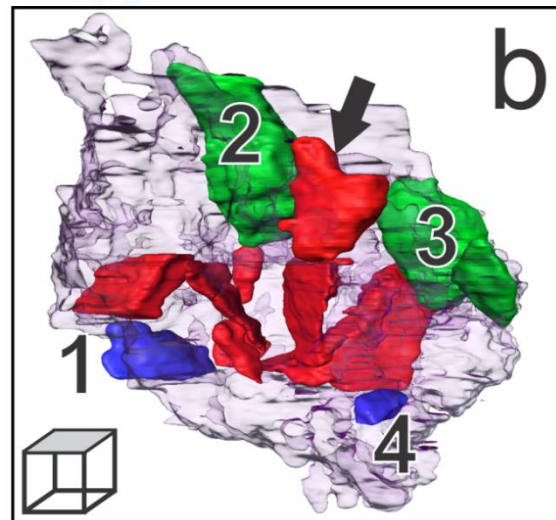
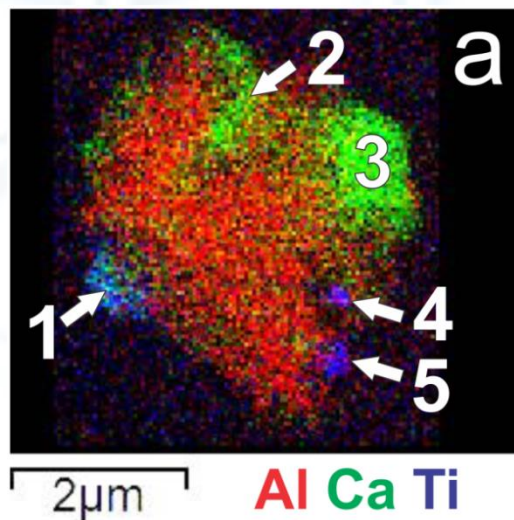
Focused Ion-Beam SEM



FIB Tomography



Particle Internal 3-D Structure

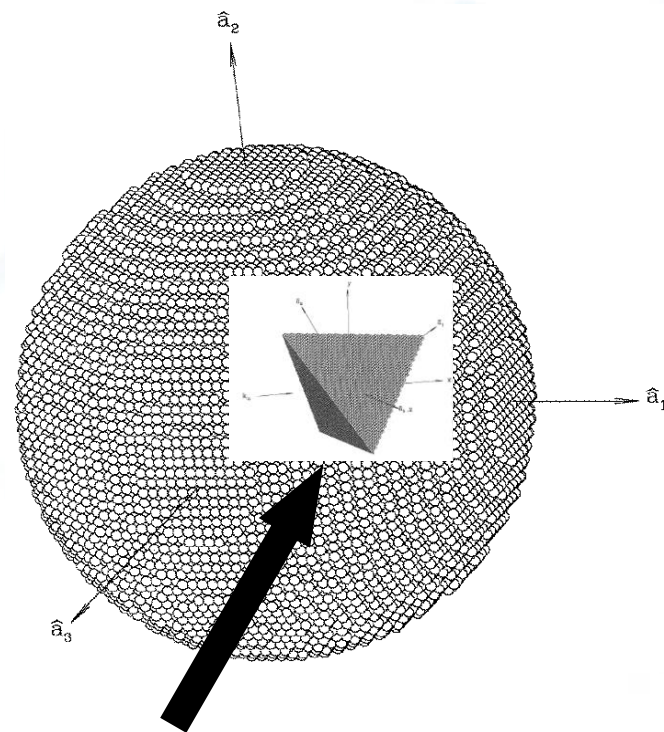


Optical Modeling: Discrete Dipole Approximation*

- Particle partitioned into M dipole 'points'
- Each dipole excited by *superposition* of external field and other dipoles' fields
- Total scattered field is numerical solution of M partial fields

Advantages

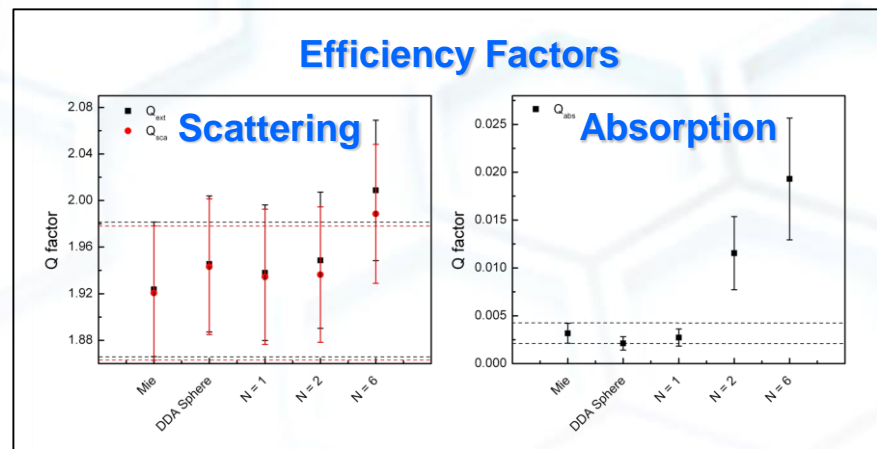
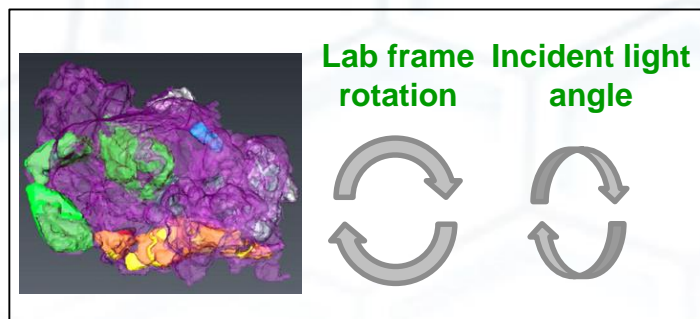
- Ideal for arbitrarily-shaped, heterogeneous, and optically-anisotropic particles
- Physical model relatively simple
- Code availability and documentation: DDSCAT 6.0 (Princeton Univ.)



Must specify dipole locations and compositions (dielectric constant or refractive index).

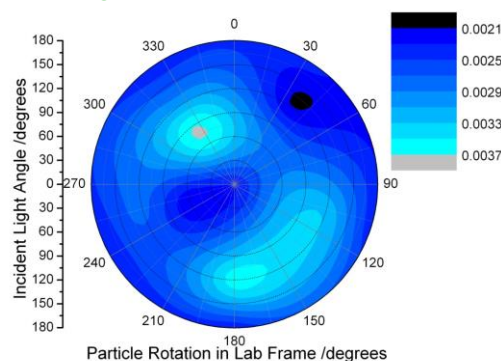
*Draine, B.T and P.J. Flatau, 1994, *J. Opt. Soc. Am*, 11, 1491-1499.

Optical Model From Particle 3-D Reconstruction

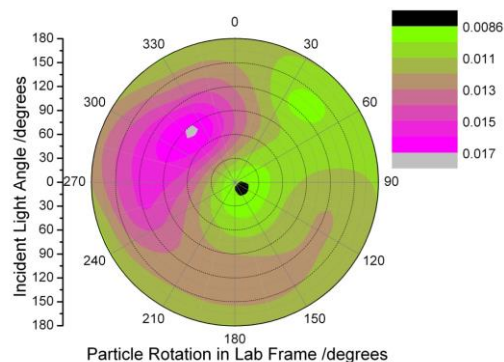


Absorption Asymmetry

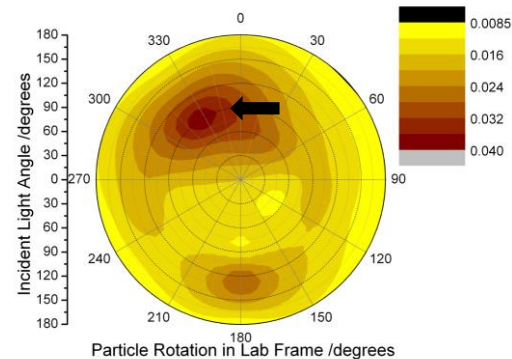
Single phase (N=1)



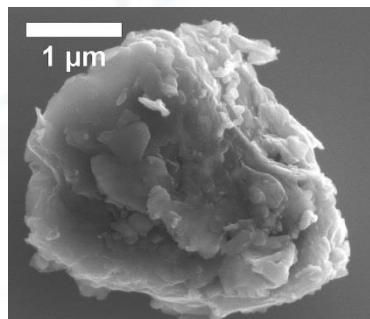
Binary phase (N=2)



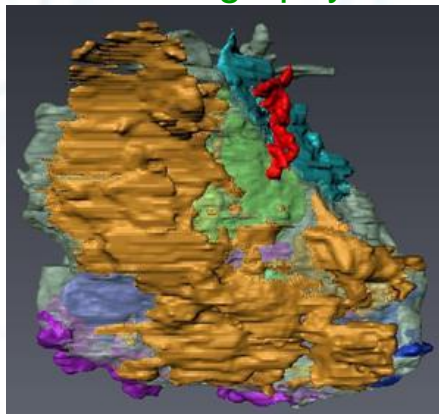
“All” phases (N=6)



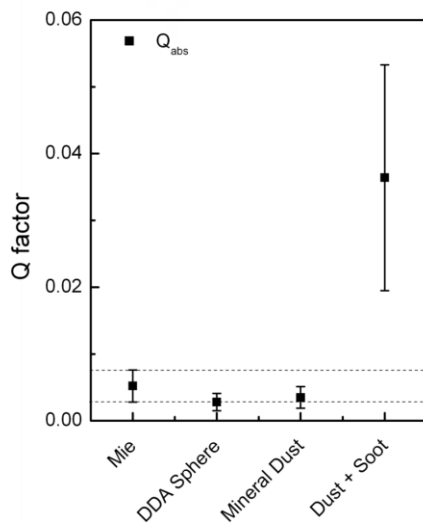
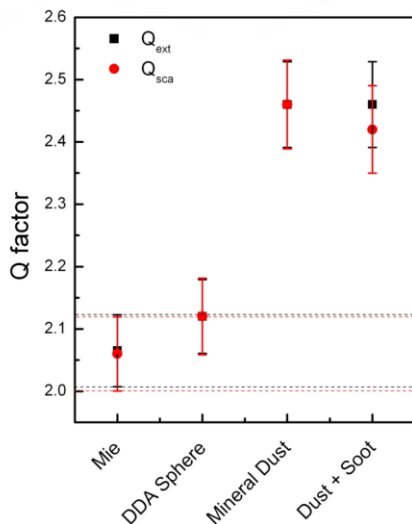
Effect of a Soot on a Mineral Particle



FIB Tomography/3D

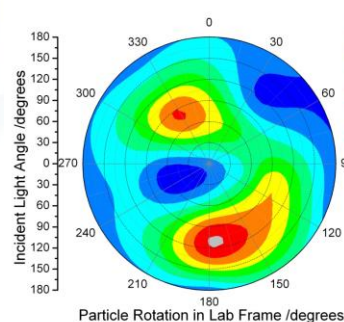


Scattering/Absorption Efficiency

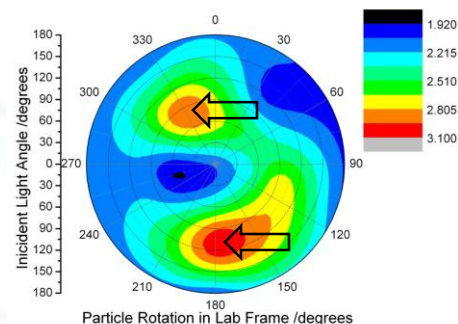


Scattering Asymmetry

Mineral Dust

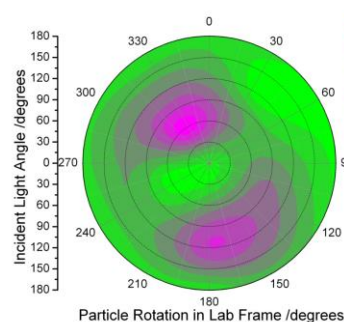


Mineral Dust + Soot

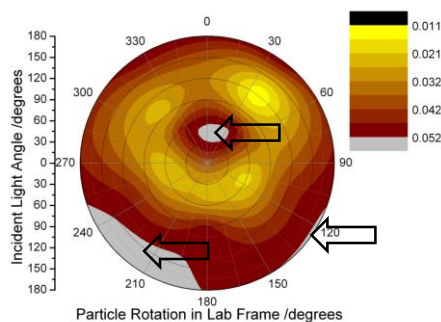


Absorption Asymmetry

Mineral Dust



Mineral Dust + Soot



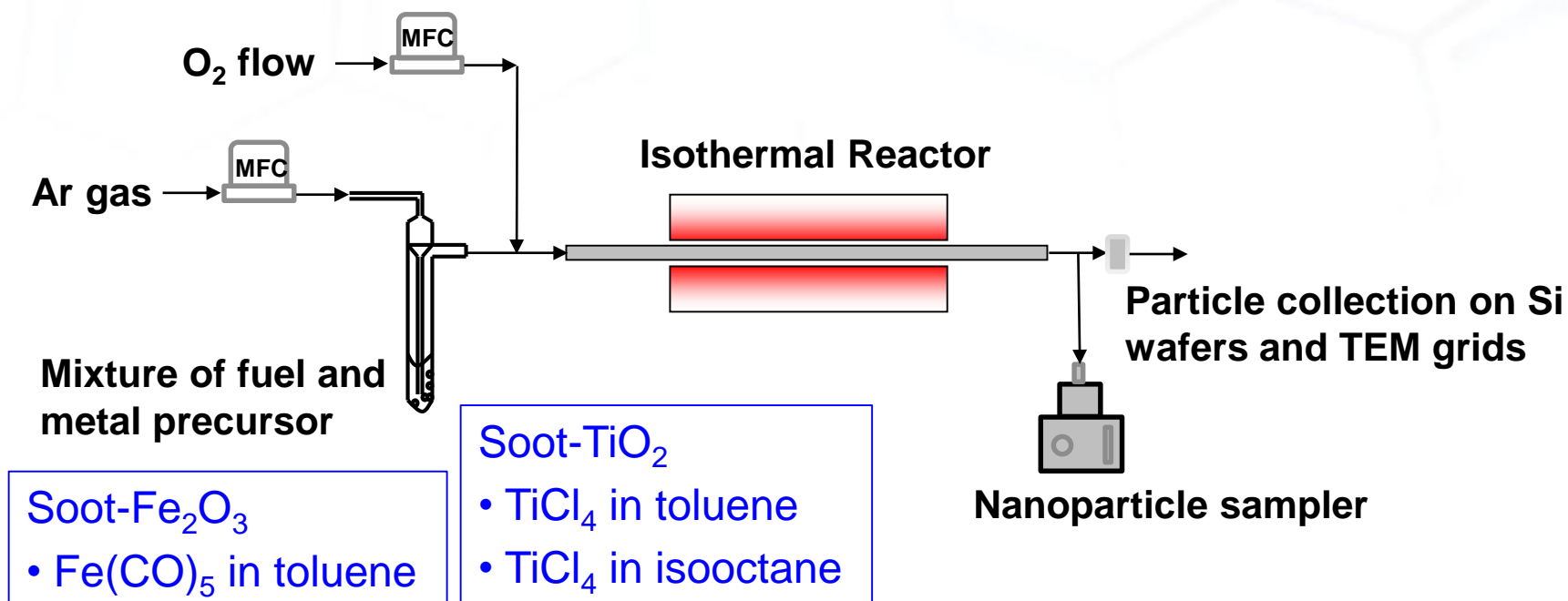
Lab-Generated Mixed-Phase Particles

Purpose

- Quantify variation in absorption and scattering in ambient particles based on lab “proxies”
- Reference materials for calibrating optical measurements

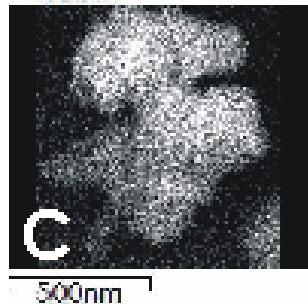
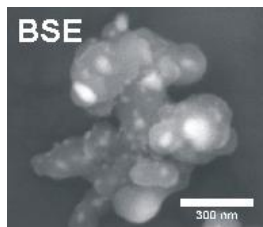
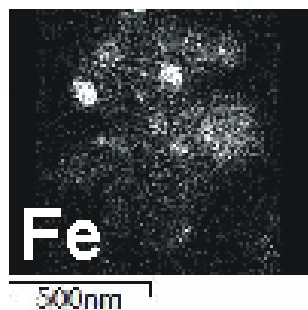
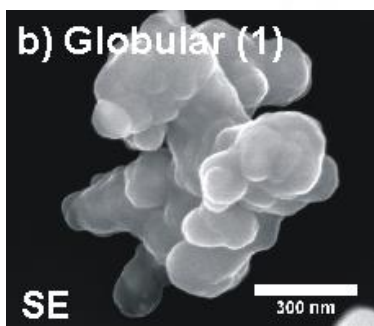
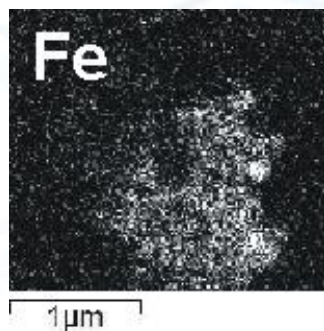
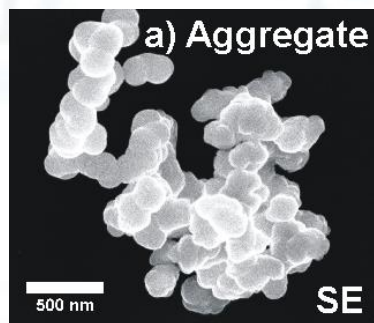
Experimental Goals

- Separate absorbing and scattering phases
- Control particle size, shape/morphology, and phase placement



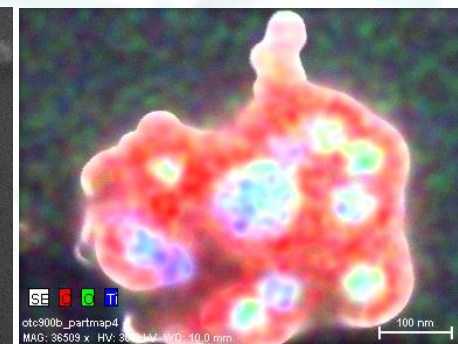
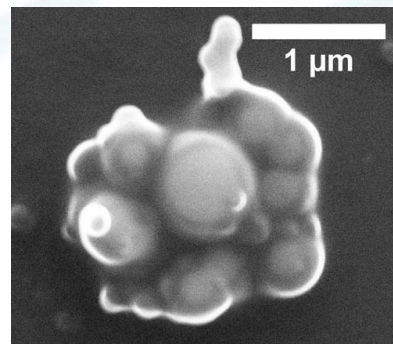
Soot-Fe₂O₃

Fe(CO)₅ in Toluene, >1000 °C

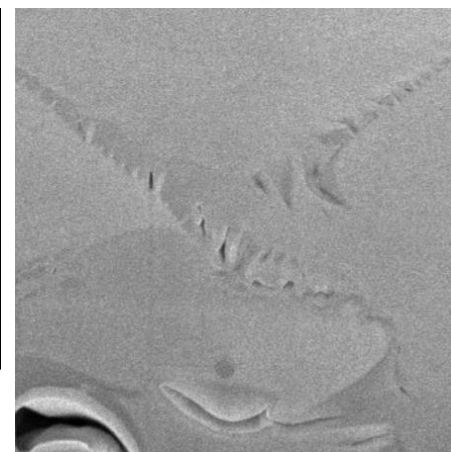
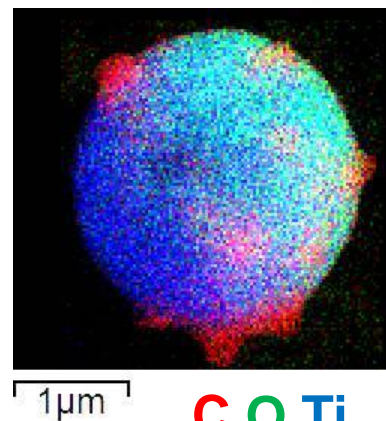


Soot-TiO₂

TiCl₄ in Isooctane, 900 °C



TiCl₄ in Toluene, 900 °C



Paradigm for Binary-Phase Particulate Reference Materials

	Strongly Scattering		Weakly Scattering																						
Strongly Absorbing	Inclusion(s) + Matrix	Candidate: Soot + TiO ₂	Inclusion(s) + Matrix	Candidate: Fe ₂ O ₃ + Soot																					
	Core / Shell	Candidate: Soot / (NH ₄) ₂ SO ₄	Core / Shell	Candidate: Soot / PAH(e.g. fluorene)																					
	Core / Shell	Candidate: TiO ₂ / Soot (High temp TiO ₂ synthesis followed by lower temp coating with soot)	Core / Shell	Candidate: Fe ₂ O ₃ / soot (High temp Fe ₂ O ₃ synthesis followed by lower temp coating with soot)																					
Weakly Absorbing	Inclusion(s) + Matrix	Candidate: Fe ₂ O ₃ + TiO ₂	<table><tr><th></th><th>R.I.</th><th>k</th></tr><tr><td>Soot</td><td>1.95</td><td>0.79</td></tr><tr><td>TiO₂</td><td>2.85-255</td><td>~0</td></tr><tr><td>Fe₂O₃</td><td>2.3-2.18</td><td>0.15-0.07</td></tr><tr><td>(NH₄)₂SO₄</td><td>1.53</td><td>~0</td></tr><tr><td>Perylene (sub. 350-400°C)</td><td>1.7</td><td>0.26</td></tr><tr><td>Fluorene (b.p. 293°C)</td><td>1.6</td><td></td></tr></table>			R.I.	k	Soot	1.95	0.79	TiO ₂	2.85-255	~0	Fe ₂ O ₃	2.3-2.18	0.15-0.07	(NH ₄) ₂ SO ₄	1.53	~0	Perylene (sub. 350-400°C)	1.7	0.26	Fluorene (b.p. 293°C)	1.6	
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Shell / Core	Candidate: PAH(e.g. perylene) / TiO ₂																								
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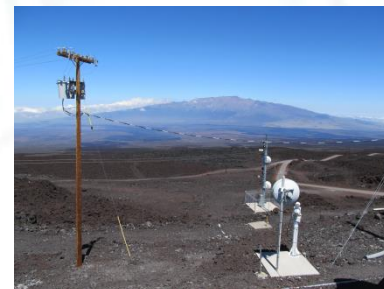
Future Directions

- Comprehensive assessment of natural variation in solar absorption and scattering for atmospheric dust:

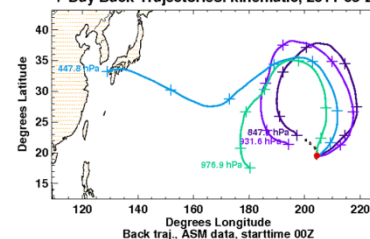
Variation assessed for

- Regions
 - Seasons
 - Urban vs. rural
 - Aged aerosol**
- Suite of lab-generated particles with mixed absorption and scattering phases for optical measurement calibration
 - Size resolved
 - Morphology controlled

EPA Collaboration on Asian Dust



Starting Location Station (red dot): Mauna Loa
7-Day Back-Trajectories: kinematic, 2011-03-25



Thank you for your attention!